

E. Radiopharmaceuticals

The Production of Aqueous Solution of ^{18}F for Injection for the Bone Scanning

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$^{99\text{m}}\text{Tc}$ -labelled compounds have been widely used for bone scanning. On the other hand, Na^{18}F began to be used again for the diagnosis of bone with development of positron camera.

The $^{16}\text{O}(\alpha, \text{pn})^{18}\text{F}$ and $^{16}\text{O}(^3\text{He}, \text{p})^{18}\text{F}$ reactions in a distilled water are very useful for the production of ^{18}F aqueous solution.

With these reactions, ^7Be , ^{11}C and ^{48}V are produced as by-products. H_2O_2 is also produced in the solution by the radiolysis of water. Therefore it is necessary to purify the irradiated water and to obtain a pyrogen-free solution of Na^{18}F without by-products for clinical diagnosis. ^{18}F was produced by the $^{16}\text{O}(\alpha, \text{pn})^{18}\text{F}$ reaction by bombarding distilled water with 60 MeV α particles.

A glass vessel was designed specially for monitoring the target water level and attached on the top of the target box without the Pt-Pd reforming catalyst. The irradiated solutions were introduced into the distillation vessel through a teflon tube

by the pressure of a He gas. H_2O_2 and $^{11}\text{CO}_2$ generated in the solution during the irradiation were removed out by heating.

^{18}F was distilled from the solution, after the addition of H_3PO_4 (0.5 ml). A small amount of water (1–2 ml) was introduced into the distillation vessel and distilled again.

Pure ^{18}F solution without impurities was obtained at the yield of 90% by these procedures. A calculated quantity of 9% NaCl (pyrogen-free solution) was added into the solution to make it isotonic.

All the final solutions produced by this method could pass the pyrogen test (limulus test and rabbit test).

For example 151 mCi of ^{18}F could be obtained in the final solution at $15\mu\text{A}$ and at the irradiation time of 88 min.

These ^{18}F solutions have been used for bone scanning at the NIRS-hospital.

Synthesis of 21-fluoroprogesterone- ^{18}F and Its Distribution in Mice

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In our studies on development of assay of the target tissue in vivo by using the specific binding of γ -radio nuclide labeled hormone with receptor protein, we have labeled 21-fluoroprogesterone with ^{18}F .

For this purpose, the labeled hormone is required to have high specific activity. 21-Fluoroprogesterone was synthesized in 1956 by the reaction of 21-iodo derivative with AgF , but this method would be unadaptable because high specific activity might not be expected. Therefore, we prepared 21-fluoroprogesterone- ^{18}F from 21-hydroxypregn-4-ene-3, 20-dione methanesulfonate, K^{18}F and crown-ether (18-Crown-6). The following

is general method: K^{18}F -quartz sand was labeled with high specific activity by dry up of ^{18}F -water, about $10\mu\text{mol}$ of carrier KF and quartz sand in a platinum crucible. The K^{18}F , crown-ether and methanesulfonate in acetone or chloroform were refluxed for 1 hour. After column chromatography of the reaction mixture, 21-fluoroprogesterone- ^{18}F was obtained in an overall radiochemical yield about 7% of K^{18}F -quartz sand with specific activity of about 10 mCi/mg at the end of preparation.

This labeling system is regarded to be suitable for ^{18}F -monofluorination of active methyl group and high specific activity labeling approximately with carrier-free ^{18}F by reducing of carrier KF.